

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 0704-0188	
<p>Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing this collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Department of Defense, Washington Headquarters Services, Directorate for Information Operations and Reports (0704-0188), 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number. PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.</p>				
1. REPORT DATE (DD-MM-YYYY) March 2013		2. REPORT TYPE Viewgraph		3. DATES COVERED (From - To) March 2013- May 2013
4. TITLE AND SUBTITLE Enhanced Cyanate Ester Nanocomposites through Improved Nanoparticle Surface Interactions			5a. CONTRACT NUMBER In-House	
			5b. GRANT NUMBER	
			5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S) Christopher M. Sahagun, Andrew J. Guenther, Joseph M. Mabry			5d. PROJECT NUMBER	
			5e. TASK NUMBER	
			5f. WORK UNIT NUMBER Q0BG	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Air Force Research Laboratory (AFMC) AFRL/RQRP 10 E. Saturn Blvd. Edwards AFB CA 93524-7680			8. PERFORMING ORGANIZATION REPORT NO.	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES) Air Force Research Laboratory (AFMC) AFRL/RQR 5 Pollux Drive Edwards AFB CA 93524-7048			10. SPONSOR/MONITOR'S ACRONYM(S)	
			11. SPONSOR/MONITOR'S REPORT NUMBER(S) AFRL-RQ-ED-VG-2013-053	
12. DISTRIBUTION / AVAILABILITY STATEMENT Distribution A: Approved for Public Release; Distribution Unlimited. PA#13210				
13. SUPPLEMENTARY NOTES Viewgraph for the SAMPE Spring Technical Conference, Long Beach, CA, 9 May 2013.				
14. ABSTRACT This work presents the results of an investigation into the role of silica nanoparticle surface chemistry in the enhancement of cyanate ester nanocomposite properties. Previous work has shown that the incorporation of silica nanoparticles improves the thermo-oxidative qualities, moisture uptake properties and processability of cyanate ester resins. This work seeks to better understand the root of these improvements by comparing thermal and mechanical properties as a function of nanoparticle loading and surface treatment using both polar and non-polar as well as potentially reactive surface functionalities. This presentation will focus on the structure-process-property relationships of various silica nanoparticle/cyanate ester systems with a discussion of the implications for novel composite processing techniques.				
15. SUBJECT TERMS				
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT SAR	18. NUMBER OF PAGES 22
a. REPORT Unclassified	b. ABSTRACT Unclassified	c. THIS PAGE Unclassified		
				19a. NAME OF RESPONSIBLE PERSON Joseph Mabry
				19b. TELEPHONE NO (include area code) 661-525-5857



Enhanced Cyanate Ester Nanocomposites through Improved Nanoparticle Surface Interactions



**Christopher Sahagun^{*1},
Andrew Guenther², Joseph Mabry²**

¹National Research Council / Air Force Research Laboratory, Edwards AFB, CA

²Air Force Research Laboratory, Aerospace Systems Directorate, Edwards AFB, CA

^{*}(661) 275-5093 christopher.sahagun.ctr@edwards.af.mil

Distribution A: Approved for public release; distribution is unlimited.

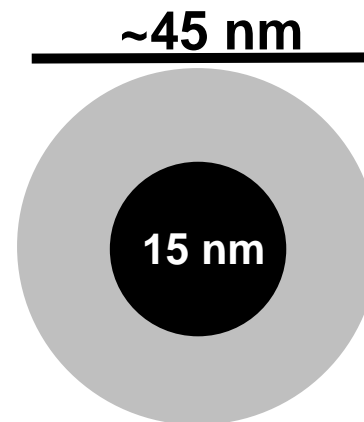


Novel Processing of an Iron(II,III)Oxide/Cyanate Ester Nanocomposite

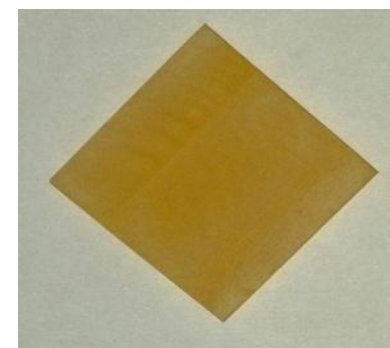
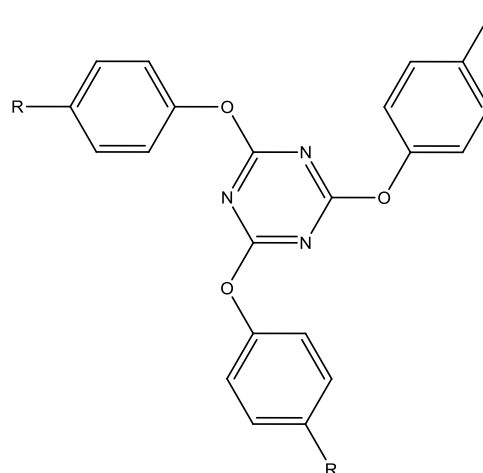
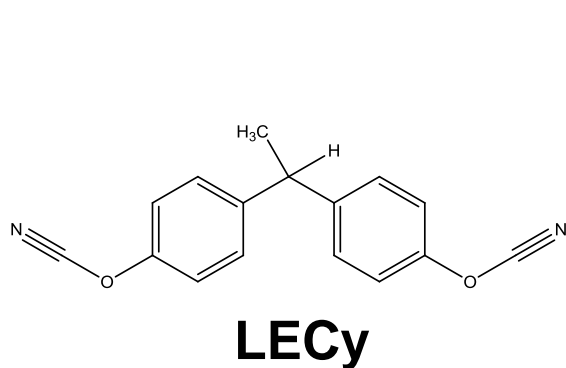


Induction Coil

Develop a high-temp nanocomposite that can be 'cured' by induction heating.



Silica-Coated Magnetite Nanoparticle

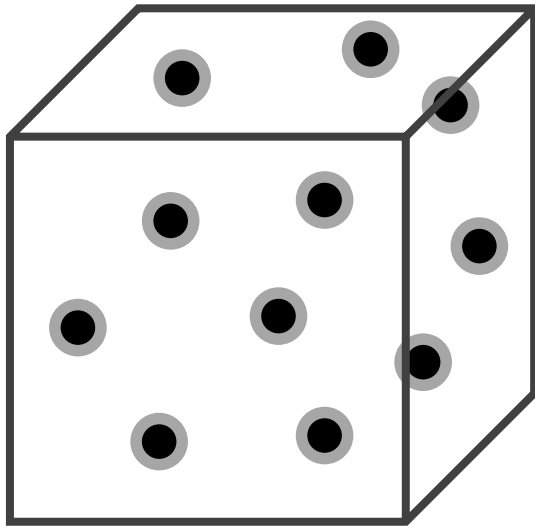


Cured LECy

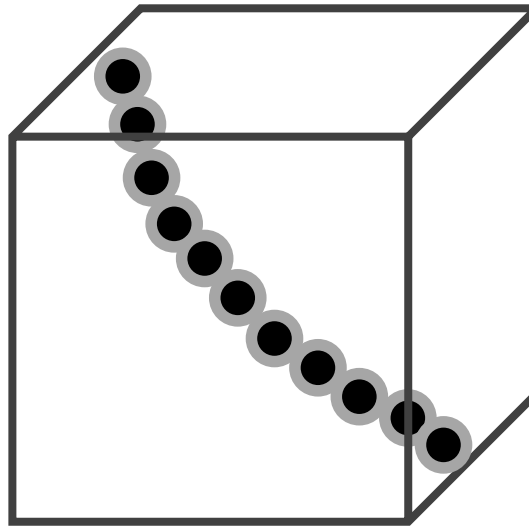


Multifunctional Structures

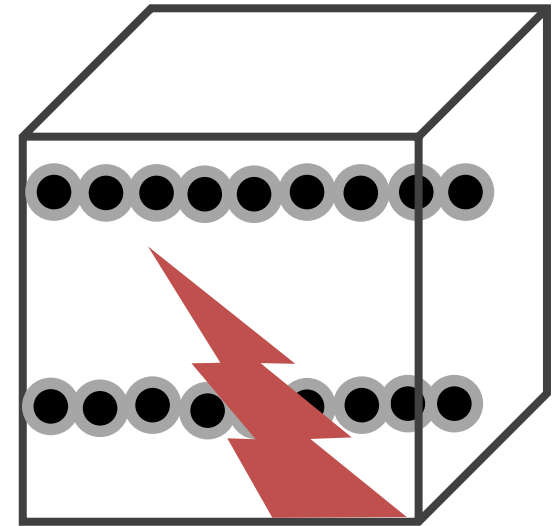
- **Controlled molecular network growth**
- **Multi-functional structures: Energy storage, damage detection, sensors, stimulus response...**



**Magnetically
Seeded Cure**



**Magnetically
Steered Positioning**



Sensing



Known Unknowns

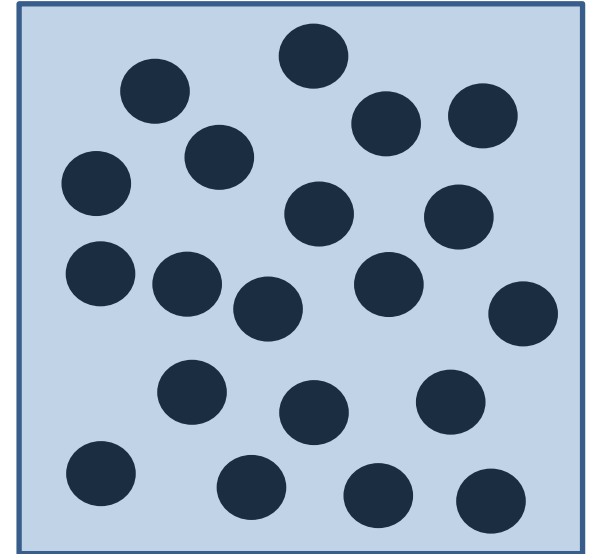
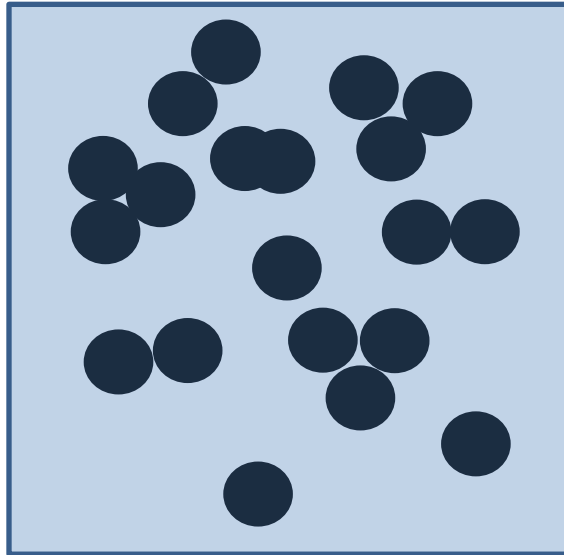
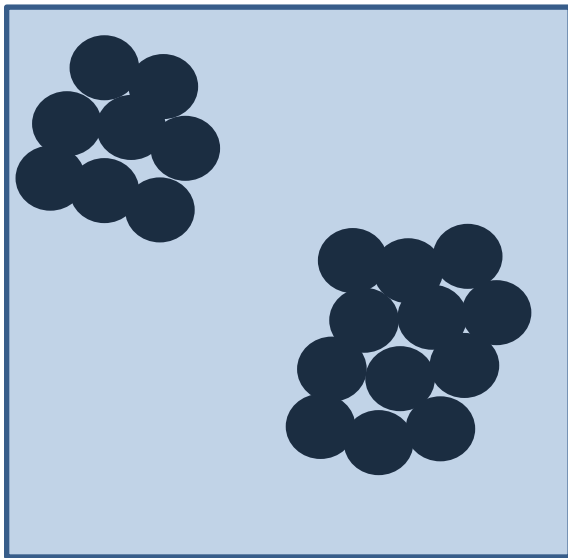


- **Particle Dispersion**
- **Resin Cure**
 - Is there a catalytic effect?
 - Does this affect the reaction pathway?
- **Heating**
 - Is it sufficient to drive the reaction?
- **Surface Modification**



The Magnetic Nanoparticles

- **Magnetite:** Iron(II,III)Oxide ($\text{FeO} \cdot \text{Fe}_2\text{O}_3$)
- **15 nm nominal diameter**
- **Received in water with no coating/dispersant**

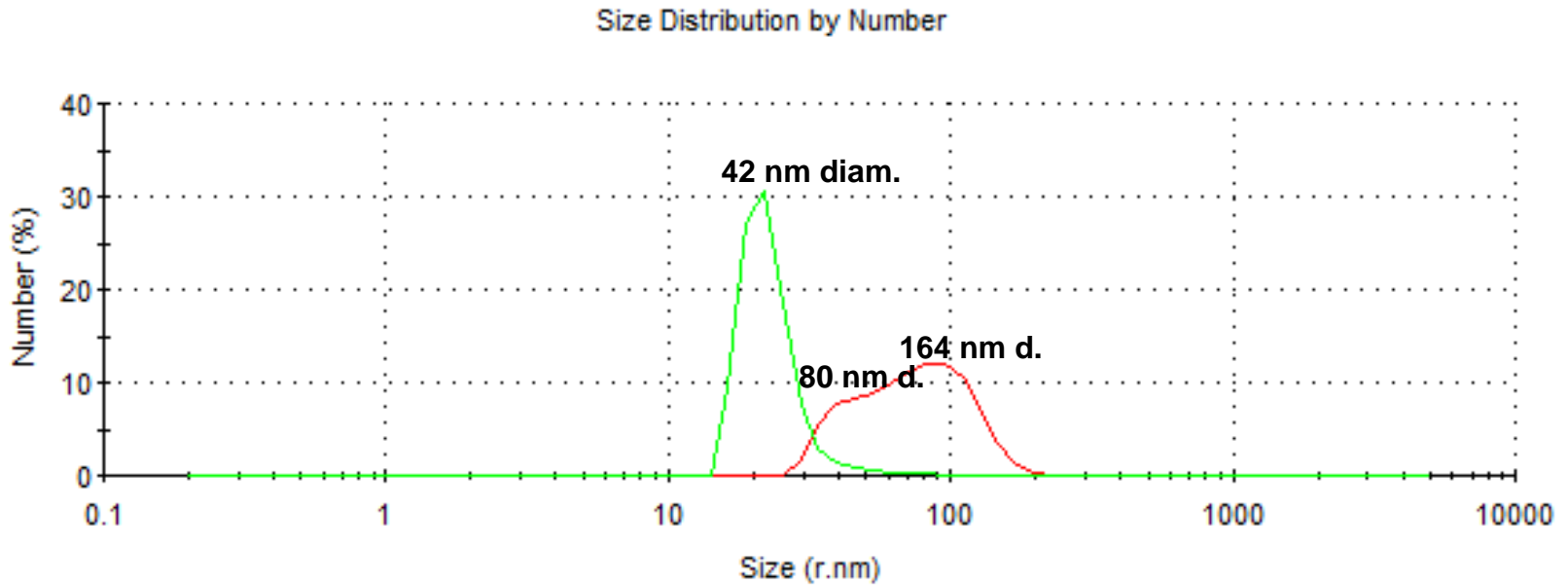


Disperse by sonication



Nanoparticle Dispersion

As received (red) vs. diluted and sonicated for 1 hour (green)



Record 44: As received 1 (averaged)

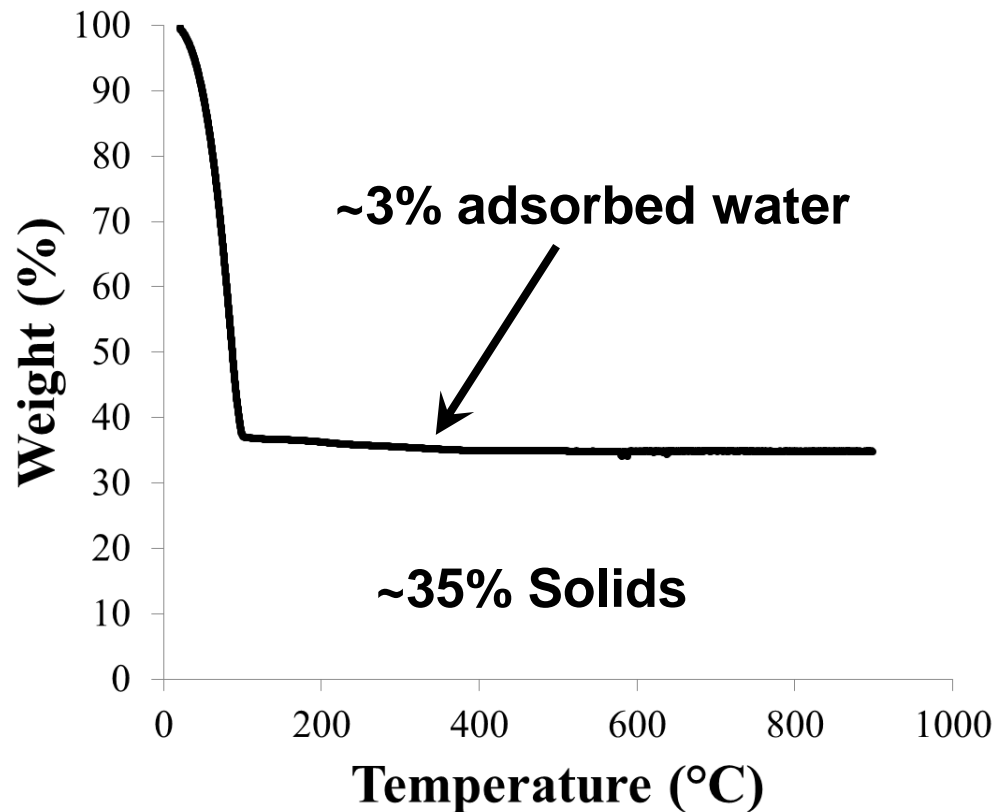
Record 45: Control - sonicated but no reaction 1 (averaged)

Sonication breaks up agglomeration, but not completely.



Water Removal

TGA of As-Received MNPs:

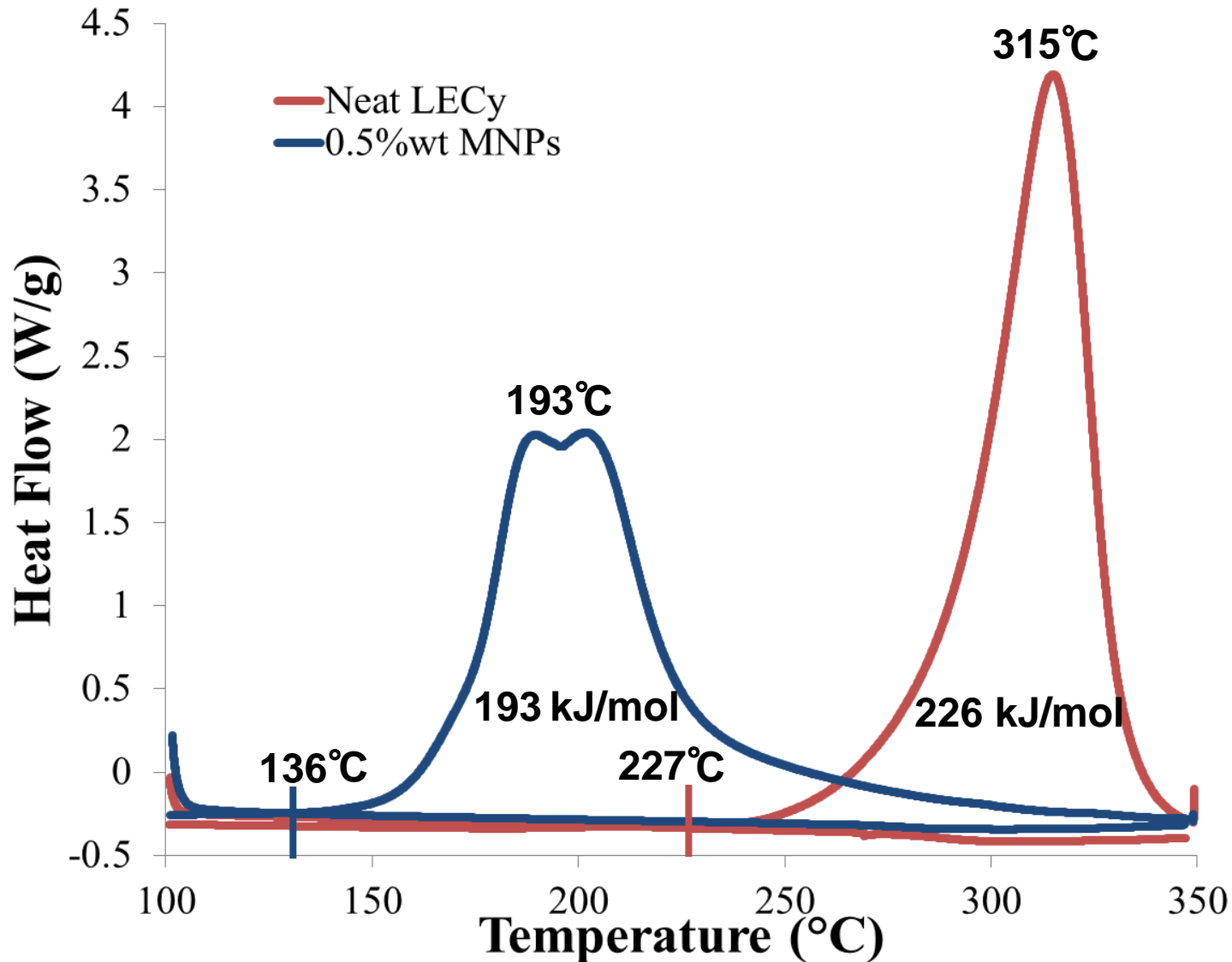


1. Add MNP solution to 20 mL acetone
2. Sonicate for 15 min, separate with magnet then decant
3. Repeat once
4. Add 20 mL dichloromethane
5. Use heat to remove dichloromethane

Dichloromethane is easy to remove



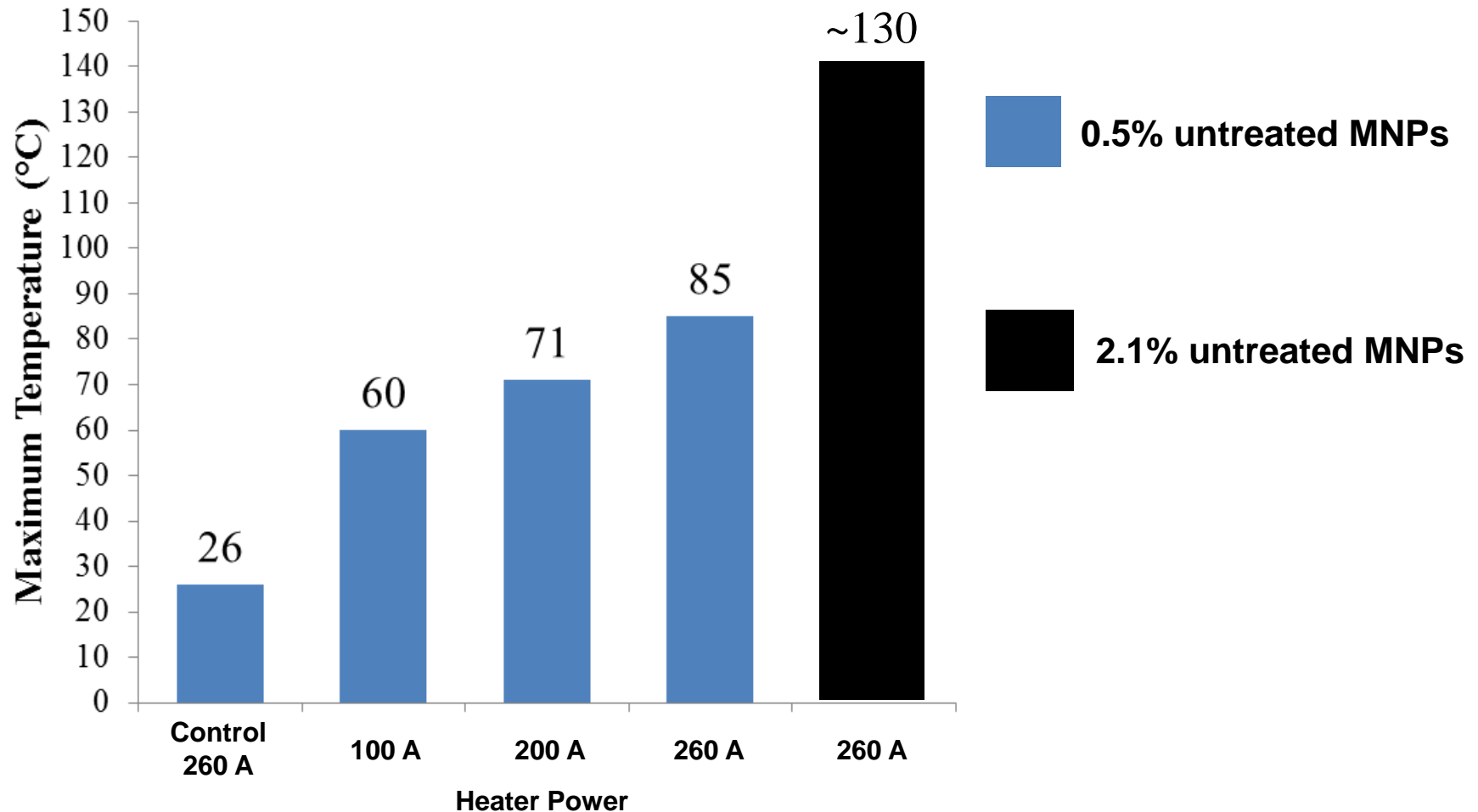
The Influence of Unmodified MNPs on Cure



Distribution A: Approved for public release; distribution is unlimited.



Induction Heating Unmodified MNPs





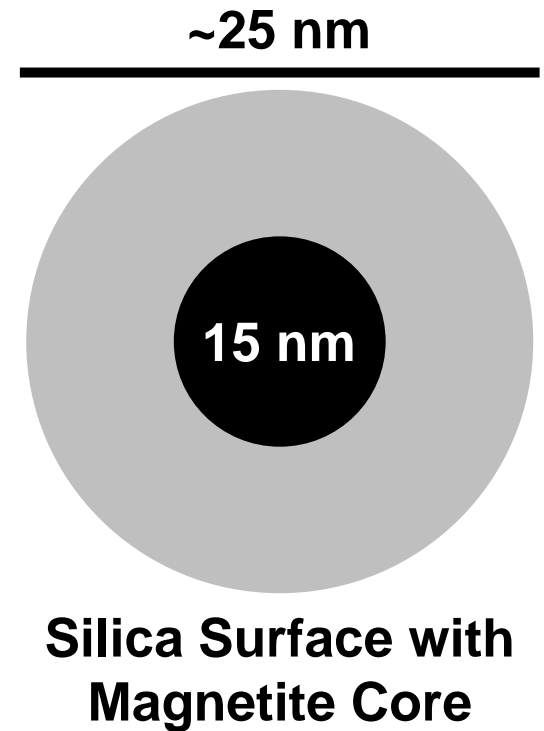
Silica-coated Magnetite Nanoparticles



Unmodified MNPs have a clear catalytic effect on the reaction...

Coat Magnetite with Silica

- Silica nanoparticles are commonly used to modify physical properties
- Iron oxides have unknown impact on curing reaction, long-term performance, etc.





Coating MNPs with Silica

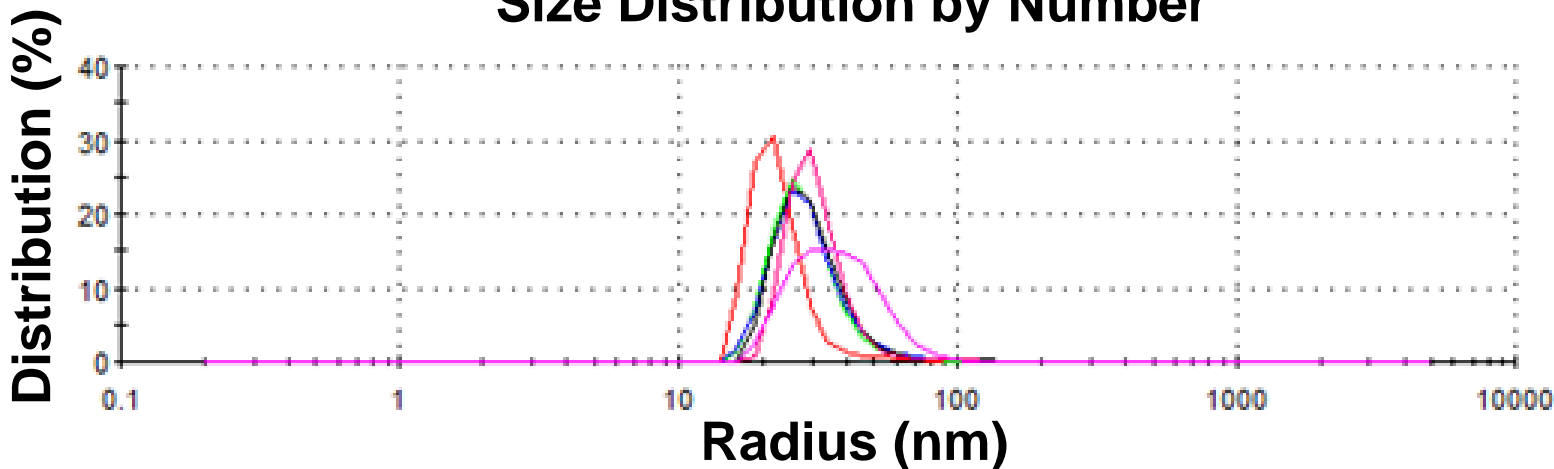


- **Modified Stöber process**
 - Use MNP as a nucleation site
- **Use amount of TEOS to control thickness**
- **Method:**
 1. Rinse 0.1 g MNPs with 0.05 M HCl
 2. Separate with magnet, decant
 3. Rinse with milli-Q water, decant (x2)
 4. Add MNPs with milli-Q water to ethanol to make an 80:20 mix
 5. Sonicate for 3 hours
 6. Add 1 mL of 30% ammonium hydroxide
 7. Slowly add 0.2639 mL TEOS
 8. Allow to react while sonicating solution. Take periodic samples for DLS.



Dynamic Light Scattering

Size Distribution by Number



Reaction Time	Diameter (nm)
0 hours	43 nm
1 hour	50 nm
2 hours	50 nm
3 hours	50 nm
4 hours	60 nm
30 hours	70 nm

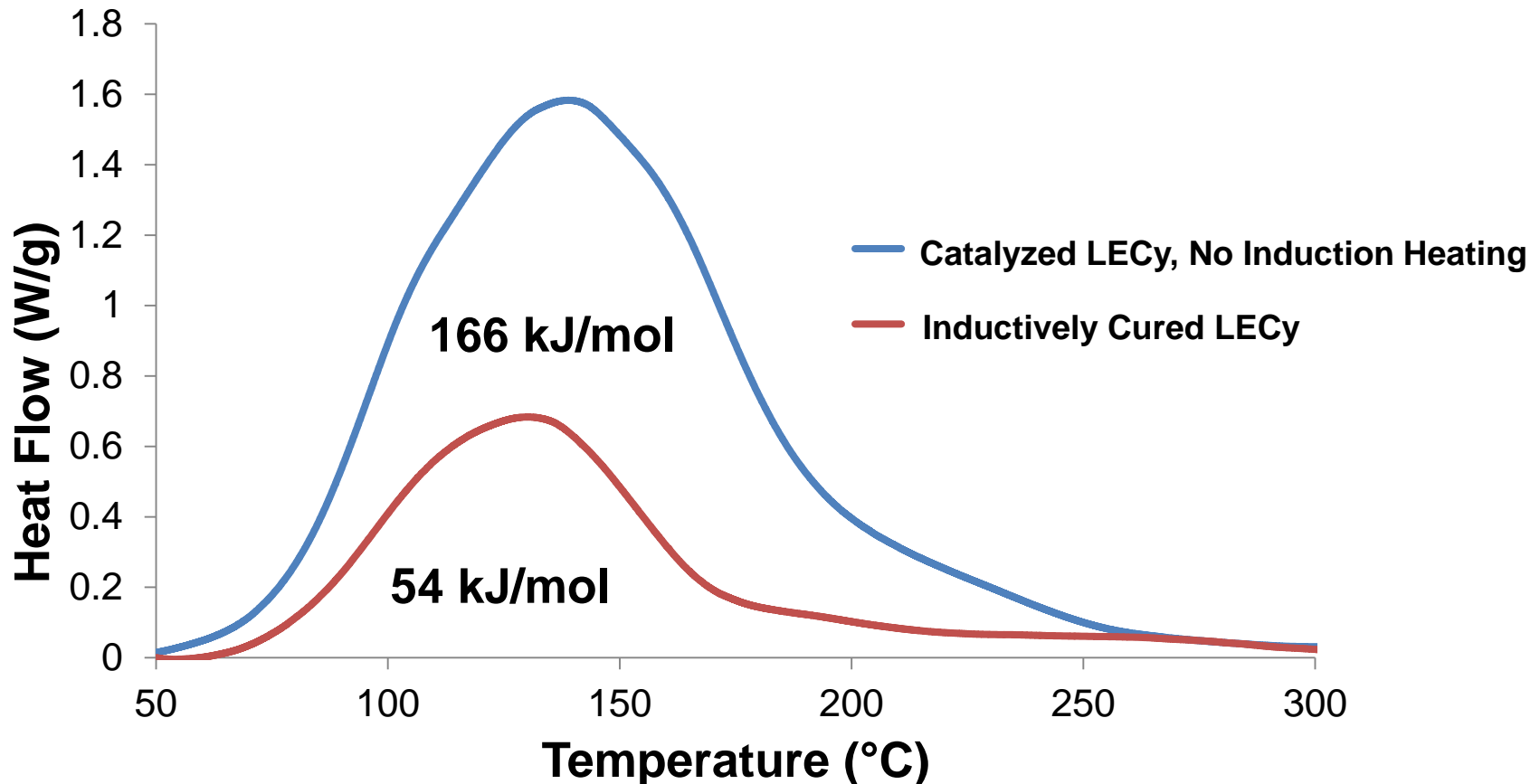
- Most reaction occurs within one hour
- Broad distribution of sizes (still nanoscale)
- No TEOS self-condensation
- Stable dispersion in EtOH/H₂O



Resin Cure via Induction Heating

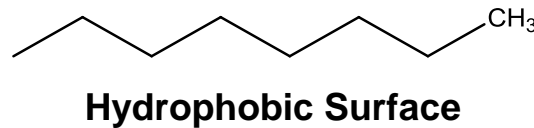
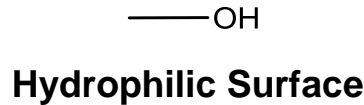
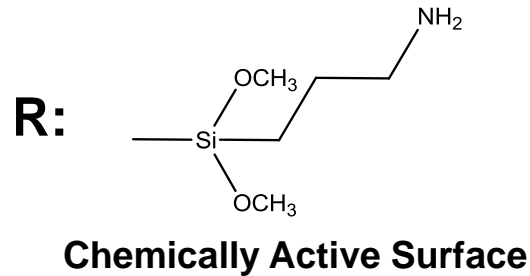
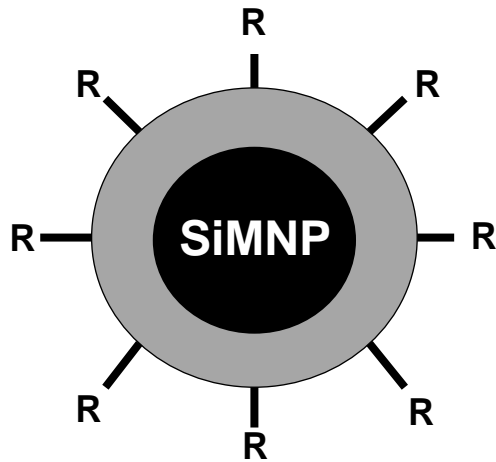


DSC scan of silica modified MNPs (NanoGAP) in LECy





Surface Modification of Silica Nanoparticles



L to R: Octyl, hydroxyl, 3-aminopropyltrimethoxy modified silica nanoparticles. (Evonik)

- **Structure/Processing/Property Relationships**

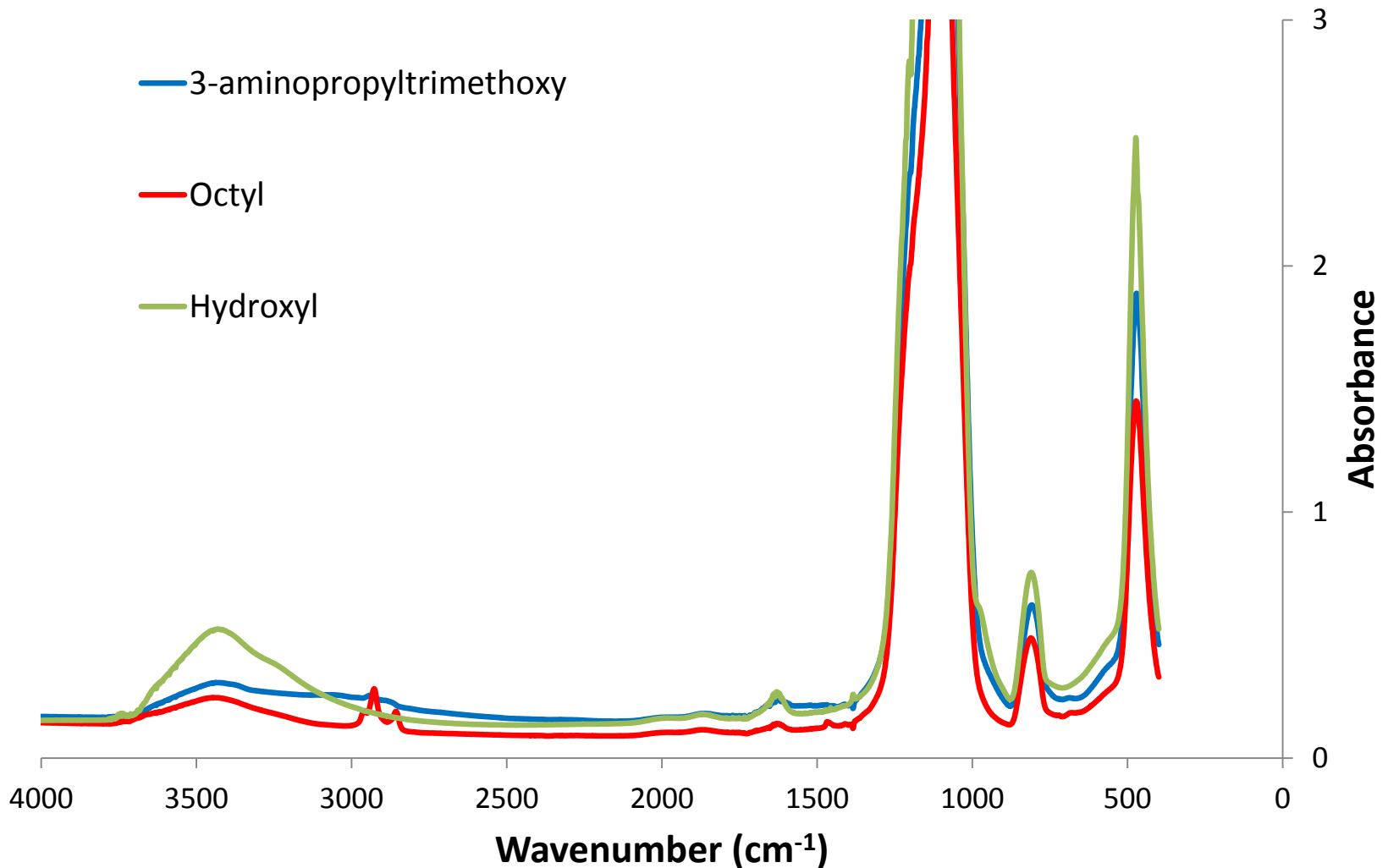
- Particle dispersion
- Reaction Kinetics
- Glass Transition Temperature
- Water uptake

Campos; Guenther; Haddad; Mabry.
Fluoroalkyl-Functionalized Silica Particles:
Synthesis, Characterization and Wetting
Characteristics. *Langmuir*. **2011**, 27,
10206-10215

DOI: 10.1021/LA201545A



IR Spectra Showing Modification of Silica Nanoparticle Surfaces

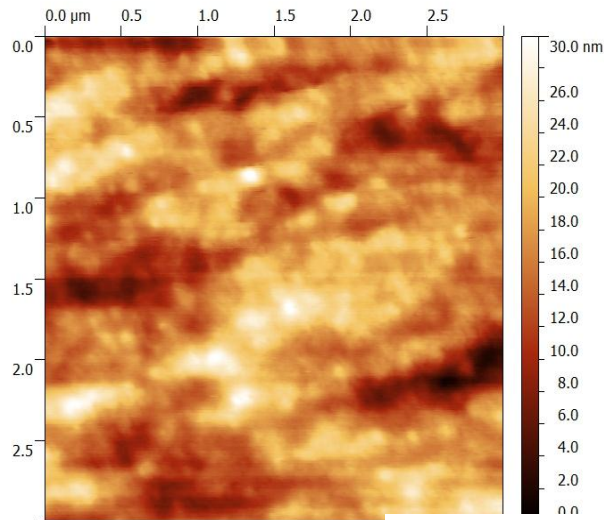




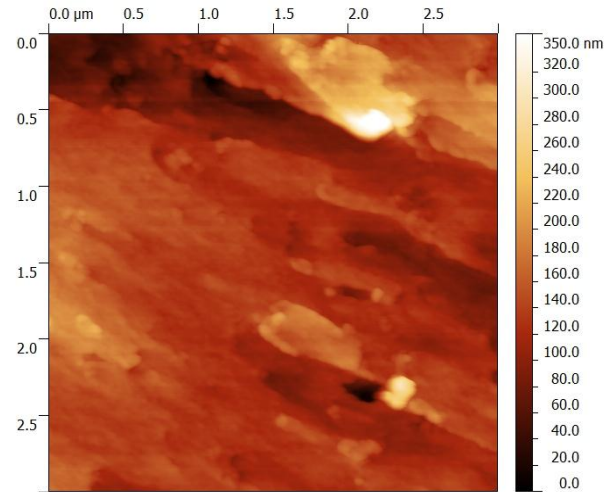
Different Surfaces Show Different Dispersion in LECy



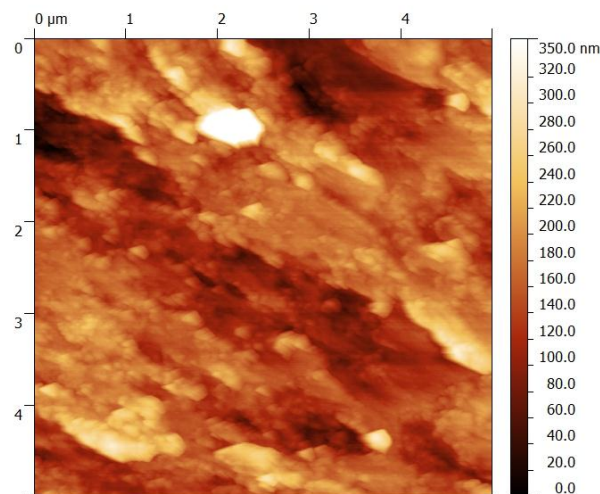
Topographic AFM Images of Sample Fracture Surface:



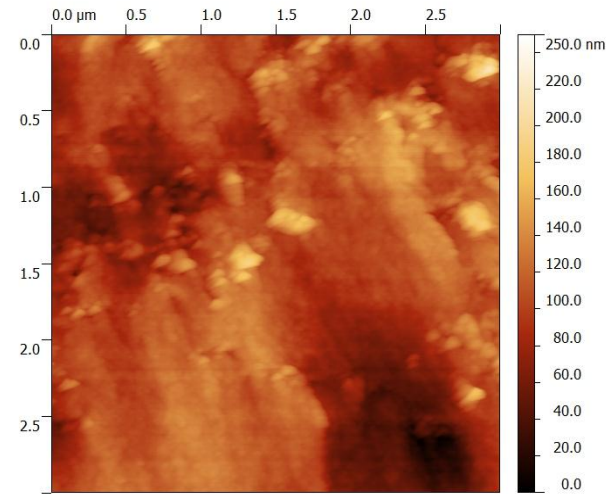
Neat LECy



3-amino



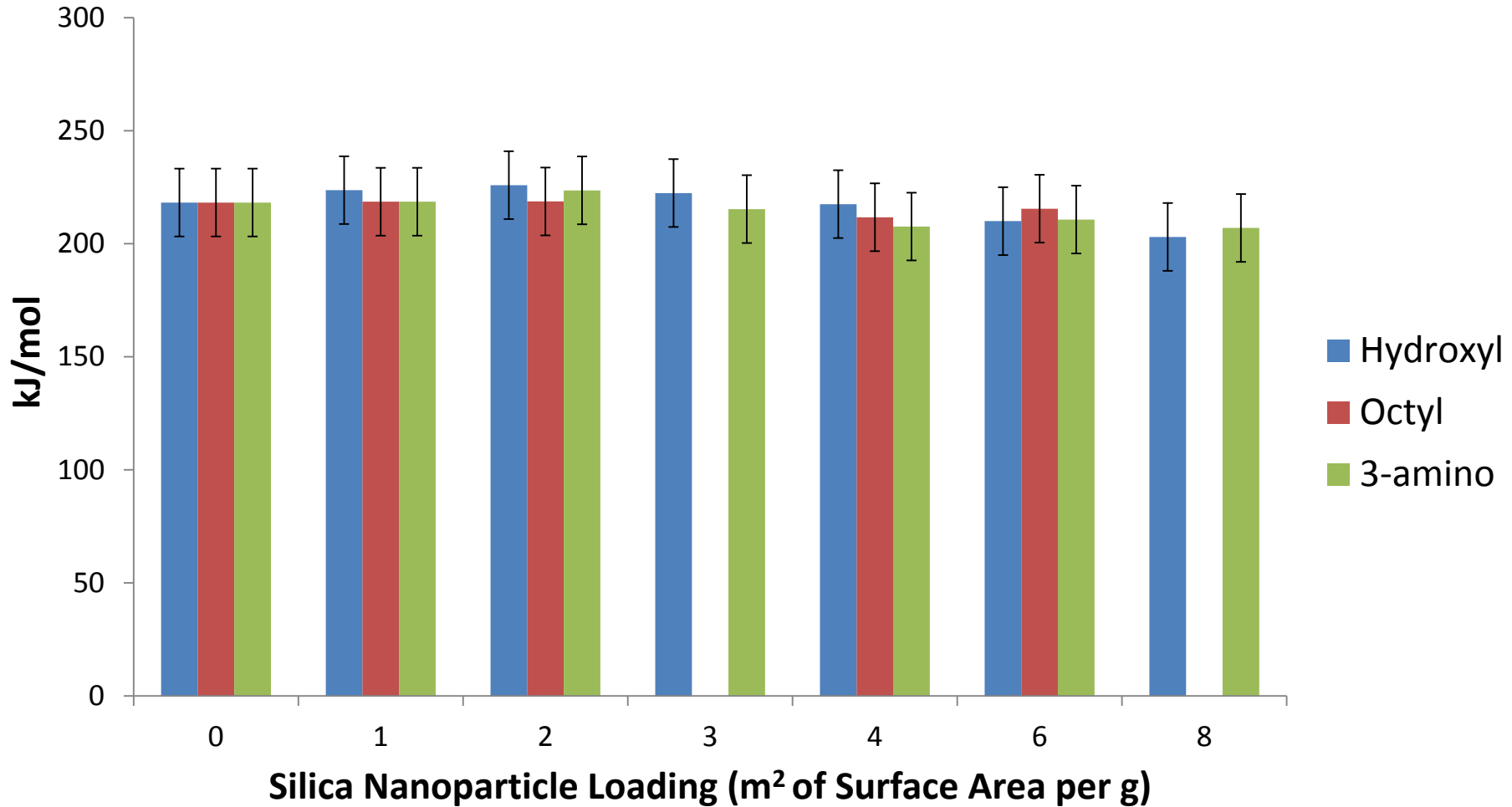
Hydroxyl



Octyl



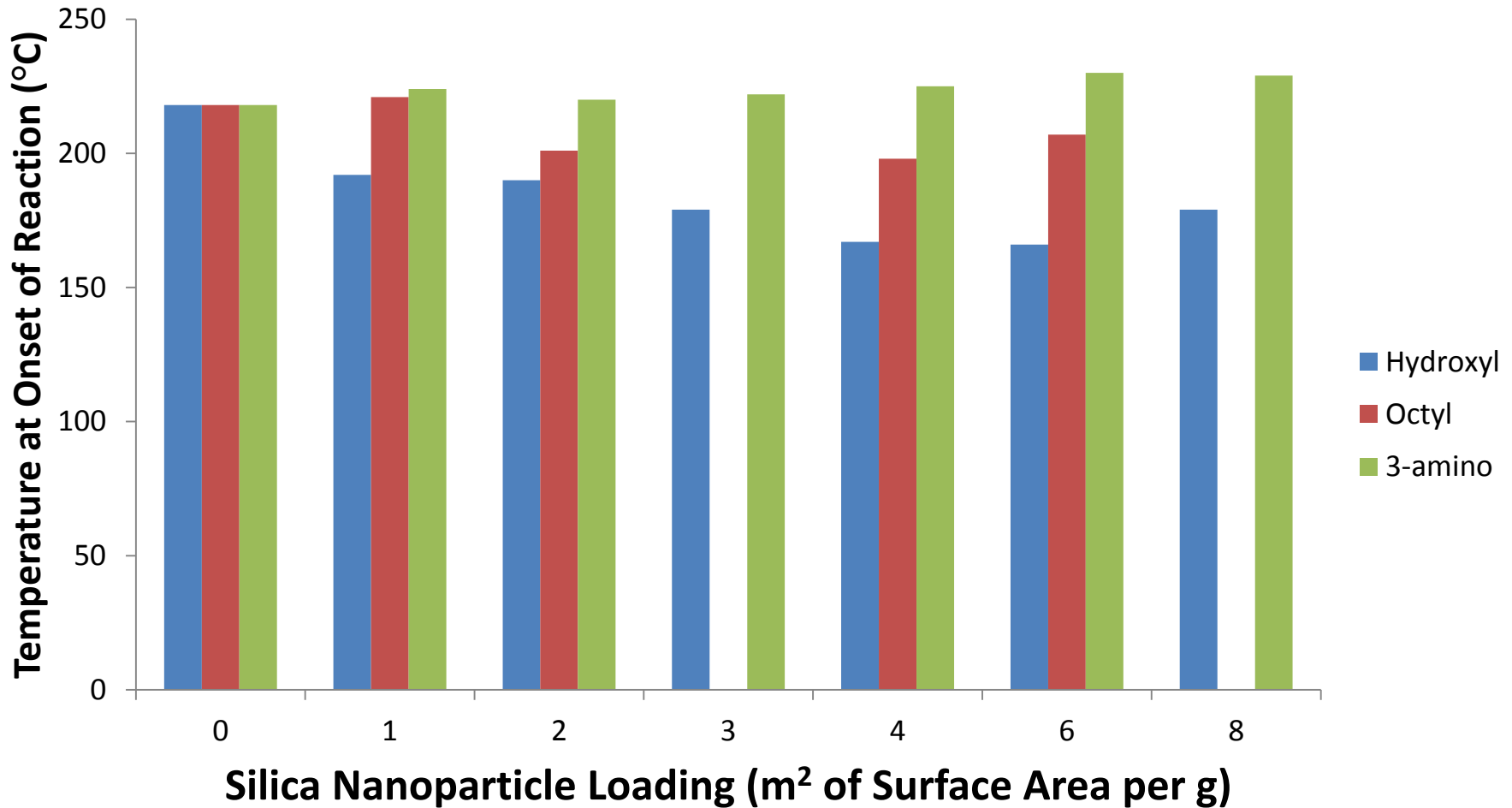
Integrated Heats of Reaction



No significant impact on reaction pathway.



Reaction Onset Temperature



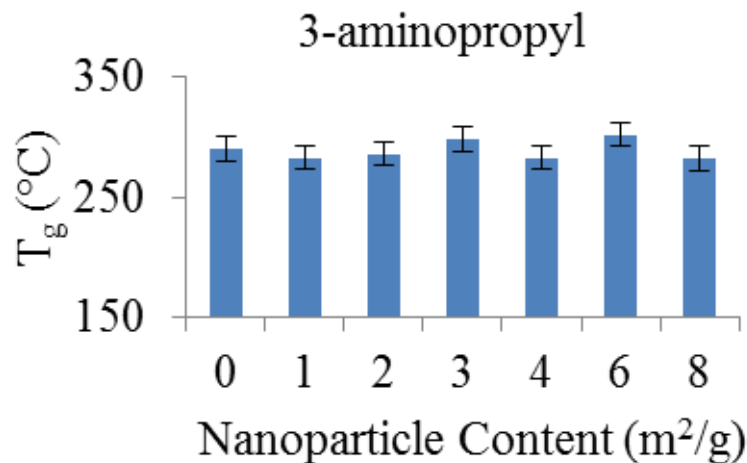
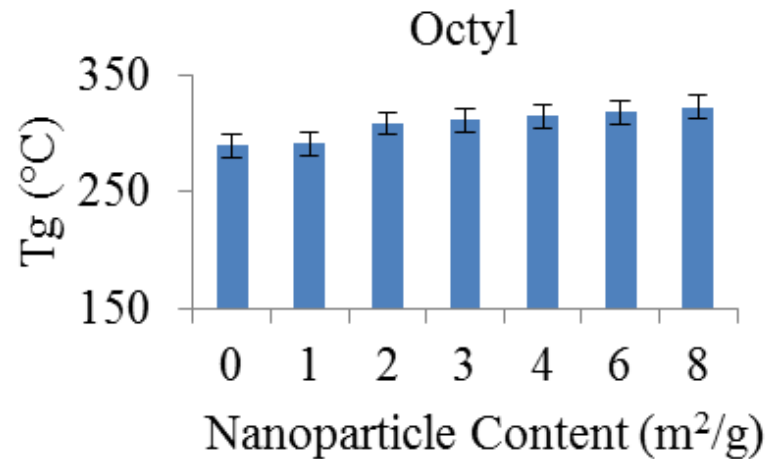
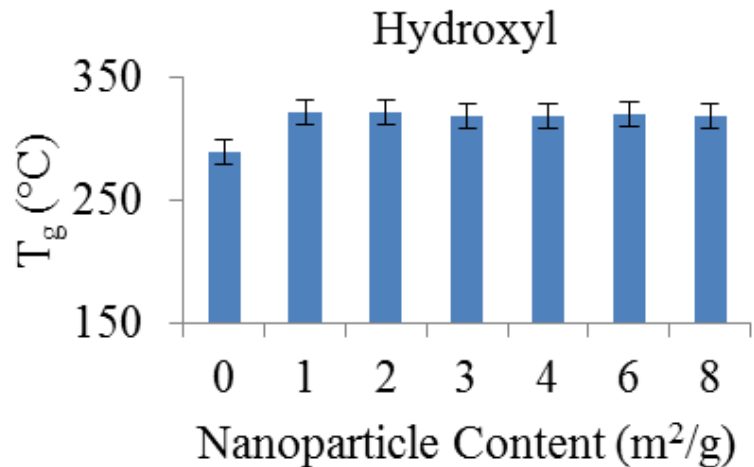
Hydroxyl catalyzes, 3-amino may delay the cure reaction



Glass Transition Temperature of Modified Silica Nanocomposites



Postcured Samples Measured by TMA:



Conversion is more important than surface chemistry



Review



- **Magnetic nanoparticle nanocomposites can enable unique processability and functionality**
- **Lab-scale induction heaters can heat low loadings to well above 100° C.**
- **Silica modification of the MNP surface is straightforward.**
- **No need to further modify the silica surface.**

Acknowledgements:

Air Force Office of Scientific Research

National Research Council

Dr. Tim Haddad (ERC)

Dr. Sean Ramirez (ERC)

Jeremiah Hubbard, Dr. Carlos Rinaldi (UPRM)

PWG team members (AFRL/RQRP)

The logo for AFRL (The Air Force Research Laboratory) features the letters 'AFRL' in a large, bold, sans-serif font. The 'A' and 'F' are grey, while the 'R' and 'L' are dark blue. To the right of the text is a stylized globe composed of a grid of small dots in blue and yellow. Below the main text, a thin blue horizontal line separates it from the tagline.

AFRL

THE AIR FORCE RESEARCH LABORATORY
LEAD | DISCOVER | DEVELOP | DELIVER